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Samarium(II)-induced ring-expansion reaction of 1,2cyclobutanedicarboxylates to produce cyclopentanones

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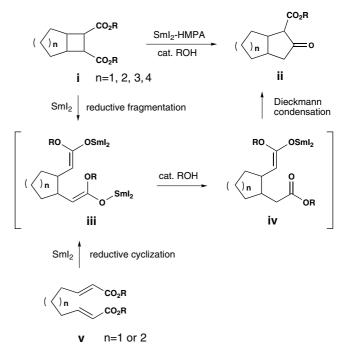
Abstract—Novel ring-expansion reaction of 1,2-cyclobutanedicarboxylates with Sm(II) in the presence of HMPA with a catalytic amount of methanol was found to provide 2-oxocyclopentanecarboxylates.

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Cyclopentanones are structural constituents present in numerous natural products, which serve to make up a polycyclic framework or isolated ring. Tandem reductive coupling-Dieckmann condensation of bis- α , β unsaturated ester v was recently noted to produce bicyclic cyclopentanones ii (n=1 or 2), a process induced by samarium(II)iodide (SmI2) via iii and iv (Scheme 1).¹ The formation of oxocyclopentanecarboxylates ii (n=3, 4) failed to occur by the above cyclization, though it was considered that reductive ring cleavage of the cyclobutane ring activated by a 1,2bis(alkoxycarbonyl) group would possibly serve as an alternative means for generating intermediate iii to produce cyclopentanones ii bearing carbocycles of various sizes. The present paper describes novel ringexpansion reaction of 1,2-cyclobutanedicarboxylates i to afford ii (n = 1-4) via Sm(II)-induced tandem reductive fragmentation-Dieckmann condensation.²

Ring transformation precursors 2, 3, 4 and 5 were prepared from the corresponding cycloalkenes 1 (n = 1-4)in three steps,³ (1) photocycloaddition with maleic anhydride,⁴ (2) methanolysis and (3) methylation, as shown in Scheme 2. Precursors 6^5 , 7^6 and 8^7 were prepared according to the literature.

The results of reactions of dimethyl 1,2-cyclobutanedicarboxylates 2-8 with Sm(II) are summarized in Table 1. On treating cyclobutanedicarboxylate 2 possessing a *cis*





ring juncture with SmI₂ (3 equiv), HMPA (12 equiv) and methanol in trace amount in THF at 0 °C, methyl 3oxobicyclo[3.3.0]octane-2-carboxylate $9a^8$ was obtained as a single isomer in 44% yield (entry 1). Higher reaction temperature (50 °C) gave a mixture of 9a (45%) and unexpected $9b^{9,10}$ (12%) possessing a *trans* ring juncture, but the total yield of 9 was noted to have increased (entry 2). HMPA addition is crucial for this reaction and

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Table 1. Sm(II)-induced ring-expansion reaction of dimethyl 1,2-cyclobutanedicarboxylates^a

Entry	Esters	Conditions		Products	Yield ^b (ratio
		Temp (°C)	Time		of a:b)
1	H CO ₂ Me H CO ₂ Me	0°C	2 h	H CO ₂ Me	44%
2	2	50 °C	1 h	$ \begin{array}{ccc} H & H \\ 9a^{c} & & & \\ 9a & & & & \\ & & 5 \\ & & \overline{H} \\ \end{array} $	CO₂Me 2 O 57% (3.8:1) ^d
3	CO ₂ Me CO ₂ Me	50 °C	1 h	H H H H H H H H H H	CO₂Me ∕ ⊃O 71% (2.6:1) ^d
ŀ	CO ₂ Me CO ₂ Me	50 °C	3 h	$ \begin{array}{c} H \\ H \\ H \\ H \\ 11a \end{array} $ $ \begin{array}{c} H \\ H \\$,CO₂Me)=0 79% (1.6:1) ^d
	CO ₂ Me CO ₂ Me	50 °C	1.5 h	$H CO_2 Me$ H	CO₂Me → 68% (1:1.5) ^d
	CO₂N CO₂Me 6	le 50 °C	1.5 h	CO ₂ Me 0	38%
Ig	Ph, 4 CO ₂ Me Ph'' ³ CO ₂ Me	50 °C	1 h	$\begin{array}{c} CO_{2}Me \\ Ph_{1,5} \\ Ph_{14} \\ Ph_{14a} \\ Ph_{$	CO₂Me → 39% (1:1.5) ^d
h	CO ₂ Me	rt	2.5 h		41%

 $^{a}\,\text{SmI}_{2}$ (3 equiv), HMPA (12 equiv) and trace amount of MeOH were used.

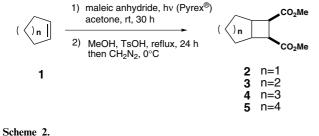
^b Isolated yield.

^cThis compound has the enol form. ^dCompounds **10a/10b** were inseparable. Compounds **9a/9b**, **11a/11b**, **12a/12b** and **14a/14b** could be separated. ^eA mixture of diastereomers was used.

^fDetermined by ¹H NMR (300 MHz) analysis.

^hSmBr₂ (5 equiv), HMPA (20 equiv) and trace amount of MeOH were used.

 $^{^{}g}\,SmI_{2}$ (5 equiv), HMPA (20 equiv) and trace amount of MeOH were used.



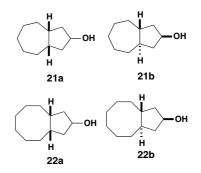
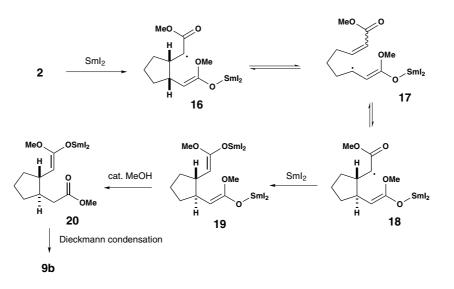


Figure 1.

recyclization of **17** to produce radical **18**, the *trans* isomer of **16**, as shown in Scheme 3.

In a typical experiment, a suspension of Sm (142 mg, 0.946 mmol) and 1,2-diiodoethane (243 mg, 0.861 mmol) in THF (5.7 mL) was sonicated for 1 h at room temperature under argon. Following addition of HMPA (0.60 mL, 3.4 mmol) and a solution of cyclobutanedicarboxylate **4** (69 mg, 0.287 mmol) and methanol (37 μ g) in THF (2.5 mL), the mixture was stirred for 3 h at 50 °C and the reaction was terminated with few drops of 30% hydrogen peroxide¹⁴ and 1 N HCl. The system was then diluted with ether, washed with saturated NaHCO₃, saturated Na₂S₂O₃ and water, dried and concentrated. The crude product was purified by silica gel column chromatography to give keto ester **11a** (29 mg, 48% yield) and **11b** (19 mg, 31% yield).

In the present study, a method was established to bring about novel ring-expansion reaction of 1,2-cyclobutanedicarboxylates, readily accessible from the corresponding cycloalkenes, to produce cyclopentanones, through application of the one electron transfer reagent Sm(II) in the presence of HMPA and methanol in trace amount. This paper reports the first instance of the transformation of cyclobutanes to cyclopentanes via tandem reductive fragmentation-Dieckmann condensation.



in the absence of which the reaction virtually fails to occur. SmI₂-HMPA treatment of cyclobutanes 3, 4 and 5, having cyclohexane, cycloheptane and cyclooctane rings, respectively, and each of these cyclobutanes being a diastereomeric mixture comprised of three components,¹¹ provided bicyclic cyclopentanones 10 (10a:10b = 2.6:1),⁸ 11 $(11a:11b = 1.6:1)^{9,12}$ and 12 $(12a:12b = 1:1.5)^{9,12}$ in 71, 79 and 68% yields, respectively (entries 3, 4 and 5). Neither bicyclo[5.3.0]decane 11 nor bicyclo[6.3.0]undecane 12 could be obtained by the previously reported SmI2-induced tandem cyclization of bis- α , β -unsaturated ester.¹ The tricyclo[5.2.1.0^{2,6}]decane derivative and monocyclic cyclopentanones could also be obtained by the present method. Reaction of dicarboxylate 6 with SmI₂-HMPA and catalytic amount of MeOH provided tricyclic keto ester 13^1 as a single isomer (entry 6), while that of 3,4-cis-dicarboxylate 7⁶ resulted in a mixture of 4,5cis-cyclopentanone 14a¹ and 4,5-trans-isomer 14b.¹ Sterically congested 3,3,4,4-tetrasubstituted cyclobutanedicarboxylate 8 does not react with SmI₂-HMPA, but on treating 8 with SmBr₂¹³-HMPA, cyclopentanone 15⁹ was obtained.

Unexpected isomerization at the ring junction via ringexpansion reaction of 2 (entry 2) to give 9b may be explained as due to generation of allylic radical 17 by radical fragmentation of the cyclopentane ring in 16 and

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References and notes

- 1. Shinohara, I.; Okue, M.; Yamada, Y.; Nagaoka, H. *Tetrahedron Lett.* **2003**, *44*, 4649–4652.
- SmI₂-promoted ring-expansion reactions have been reported. Fukuzawa, S.; Tsuchimoto, T. *Tetrahedron Lett.* **1995**, *36*, 5937–5938; Hasegawa, E.; Kitazume, T.; Suzuki, K.; Tosaka, E. *Tetrahedron Lett.* **1998**, *39*, 4059– 4062; Chung, S. H.; Cho, M. S.; Choi, J. Y.; Kwon, D. W.; Kim, Y. H. *Synlett* **2001**, 1266–1268; Iwaya, K.; Nakamura, M.; Hasegawa, E. *Tetrahedron Lett.* **2002**, *43*, 5067– 5070; SmI₂-promoted fragmentation reactions have been reported. Imamoto, T.; Hatajima, T.; Yoshizawa, T. *Tetrahedron Lett.* **1994**, *35*, 7805–7808; Yamashita, M.; Okuyama, K.; Ohhara, T.; Kawasaki, I.; Ohta, S. *Synlett* **1996**, 547–548.
- Cycloalkene 1 (2.0 equiv) and maleic anhydride (1.0 equiv) were used for photocycloaddition and overall yields of 2, 3, 4 and 5 were 13%, 35%, 43% and 37%, respectively, based on maleic anhydride. Compounds 2 and 3 can also be obtained by photocycloaddition of cyclopentene or cyclohexene with dimethyl maleate. De Mayo, P.; Reid, S. T.; Yip, R. W. *Can. J. Chem.* 1964, *42*, 2828–2835.
- Robson, R.; Grubb, P. W.; Barltrop, J. A. J. Chem. Soc., Abstr. 1964, 21, 2153–2164.
- Mitsudo, T.; Kokuryo, K.; Shinsugi, T.; Nakagawa, Y.; Watanabe, Y.; Takegami, Y. J. Org. Chem. 1979, 44, 4492–4496.
- 6. Nishikubo, T.; Takahashi, E.; Miyaji, T.; Iizawa, T. Bull. Chem. Soc. Jpn. 1985, 58, 3399–3400.
- 7. Goe, G. L. J. Org. Chem. 1973, 38, 4285-4288.
- Taber, D. F.; Ruckle, R. E., Jr. J. Am. Chem. Soc. 1986, 108, 7686–7693.
- 9. Compound **9b**: EI-MS (m/z, %) 182 (M⁺, 3), 94 (100); HR-MS calcd for C₁₀H₁₄O₃: 182.0943, found: 182.0936; IR v_{max} (neat) 1761, 1732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.29 (m, 2H), 1.67 (m, 1H), 1.85–2.19 (m, 6H), 2.48 (dd, J = 17.4, 6.4 Hz, 1H), 2.88 (d, J = 12.7 Hz, 1H), 3.76 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 25.5, 26.1, 26.8, 43.2, 45.8, 52.1, 52.3, 60.4, 169.9, 212.4.

Compound **11a**: EI-MS (m/z, %) 210 (M⁺, 33), 178 (100), 121 (37); HR-MS calcd for C₁₂H₁₈O₃: 210.1256, found: 210.1253; IR v_{max} (neat) 1755, 1732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.15-1.55 (m, 5H), 1.70–2.05 (m, 5H), 2.17 (dd, J = 18.5, 4.0 Hz, 1H), 2.57 (m, 1H), 2.61 (dd, J = 18.5, 9.0 Hz, 1H), 2.84 (m, 1H), 3.01 (d, J = 9.9 Hz, 1H), 3.74 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 26.1, 29.2, 31.2, 32.0, 32.1, 38.2, 43.6, 47.1, 52.4, 61.2, 169.6, 211.4.

Compound **11b**: EI-MS (m/z, %) 210 (M⁺, 82), 108 (100); HR-MS calcd for C₁₂H₁₈O₃: 210.1256, found: 210.1259; IR v_{max} (neat) 1757, 1732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.25–1.42 (m, 2H), 1.45–1.67 (m, 4H), 1.72 (m, 2H), 1.81-2.10 (m, 4H), 2.30 (qd, J = 10.6, 3.9 Hz, 1H), 2.52 (dd, J = 18.2, 7.0 Hz, 1H), 2.90 (d, J = 11.9 Hz, 1H), 3.74 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 26.6, 26.7, 26.9, 33.5, 34.2, 41.0, 47.3, 47.4, 52.4, 64.2, 169.5, 210.5. Compound **12a**: EI-MS (*m*/*z*, %) 224 (M⁺, 30), 192 (100); HR-MS calcd for C₁₃H₂₀O₃: 224.1412, found: 224.1411; IR v_{max} (neat) 1757, 1732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.35–1.83 (m, 12H), 2.15 (m, 1H), 2.41 (m, 1H), 2.50-2.75 (m, 2H), 2.93 (dd, J = 10.1, 1.3 Hz, 1H), 3.74 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 25.3, 26.3, 27.39, 27.41, 29.7, 30.2, 36.9, 44.5, 49.0, 52.5, 60.7, 169.7, 211.5. Compound **12b**: EI-MS (m/z, %) 224 $(M^+, 30)$, 192 (100); HR-MS calcd for C₁₃H₂₀O₃: 224.1412, found: 224.1411; IR v_{max} (neat) 1757, 1732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.45–1.78 (m, 10H), 1.84 (m, 2H), 1.97 (m, 1H), 2.11 (dd, J = 17.6, 11.7 Hz, 1H), 2.35 (m, 1H), 2.48 (dd, J = 17.6, 6.5 Hz, 1H), 2.94 (d, J = 11.9 Hz, 1H), 3.75 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 25.0, 26.8, 26.9, 31.7, 32.3, 39.6, 45.9, 47.5, 52.4, 64.2, 169.5, 210.0. Compound 15: EI-MS (m/z, %) 198 (M⁺, 19), 151 (100);

HR-MS calcd for $C_{11}H_{18}O_3$: 198.1256, found: 198.1253; IR v_{max} (neat) 1749, 1732 cm⁻¹ ¹H NMR (300 MHz, CDCl₃): δ 0.99 (s, 3H), 1.02 (s, 3H), 1.07 (s, 3H), 1.16 (s, 3H), 2.17 (d, J = 18.4 Hz, 1H), 2.39 (d, J = 18.4 Hz, 1H), 3.24 (s, 1H), 3.71 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 21.2, 22.60, 22.63, 25.3, 39.8, 44.8, 51.9, 52.1, 63.9, 169.4, 210.9.

- 10. Relative configuration of C1–C5 in **9b** was determined based on comparison of ¹H and ¹³C NMR spectra of **9b** with those of **9a**,⁸ ca 50% of which was found to exist in the enol form. The relative configuration of C1–C2 in **9b** was elucidated from the coupling constants of C2–H (2.88 ppm, J = 12.7 Hz) in ¹H NMR.
- 11. Compounds **3**, **4** and **5** were all diastereomeric mixtures but no determination was made of their respective amounts.
- 12. Configurations of ring junctures in 11a and 11b were found based on ¹³C NMR analysis of **21a** and **21b** derived from 11a and 11b by decarbomethoxylation with NaCl, DMSO and H₂O at 140 °C for 1.5 h¹⁵ and reduction of ketone with NaBH₄ in MeOH at 0 °C. The ¹³C NMR spectrum of 21a indicated six signals, while that of 21b, ten signals. ¹³C NMR (100 MHz, CDCl₃) of **21a**: δ 29.7, 31.5, 33.1, 40.8, 44.0, 72.9; ¹³C NMR (100 MHz, CDCl₃) of **21b**: δ 26.5, 28.0, 28.1, 33.9, 34.9, 43.5, 44.9, 45.1, 45.2, 72.7. Configurations of ring junctures in 12a and 12b were clarified in same manner as above. The ¹³C NMR spectrum of 22a indicated six signals and that of 22b, eight signals (more than six). ¹³C NMR (100 MHz, CDCl₃) of 22a: δ 25.9, 29.7, 30.6, 40.3, 45.7, 71.8; ¹³C NMR (100 MHz, CDCl₃) of **22b**: δ 24.7, 27.2, 35.6, 36.5, 41.3, 42.3, 45.5, 72.2 (Fig. 1).
- 13. Lebrun, A.; Rantze, E.; Namy, J.-L.; Kagan, H. B. New J. Chem. **1995**, *19*, 699–705.
- 14. Without hydrogen peroxide treatment, reduction of the ketone in **11** occurs to give a small amount of the corresponding cyclopentanols.
- 15. Shiao, M.-J.; Liang, D.; Ku, C.-S.; Yang, C.-H. Syn. Commun. 1988, 18, 1553–1563.